



# APPLICATION OF SULFATE REDUCING BACTERIA IN ACID MINE DRAINAGE ACTIVE REMEDIATION

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WHAT IS ACID MINE DRAINAGE?	PROBLEMS ASSOCIATED WITH AMD	AIMS	APPLICATION OF SULFATE REDUCING BACTERIA (SRB) FOR AMD REMEDIATION
<p>Acid mine drainage (AMD) is one of the main contaminating problems derived from the mining industry and metallurgy, as it is an acidic metal and sulfate-containing wastewater. It is mainly a consequence of the exposure of iron pyrite (FeS<sub>2</sub>) among other sulfide minerals to both oxygen and water during mining and processing of metal ores and coal, which triggers the accelerated oxidation of these minerals.</p> $2\text{FeS}_2 + 7\text{O}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{Fe}^{2+} + 4\text{SO}_4^{2-} + 4\text{H}^+$	<ul style="list-style-type: none"><li>» Contaminated drinking water.</li><li>» Affected growth and reproduction of plants and animals.</li><li>» Corroding effects of acid on infrastructures.</li></ul>	<p>Among all the bioremediation strategies for AMD, this work focuses on active technologies. The two principal standards will be described and compared. Therefore, real examples will be analyzed.</p>	<p>Under anaerobic conditions, SRB catalyze the reduction of sulfate to sulfide, generating alkalinity by transforming a strong acid (sulfuric) into a relatively weak acid (hydrogen sulfide):</p> $\text{SO}_4^{2-} + 2\text{CH}_2\text{O} + 2\text{H}^+ \rightarrow \text{H}_2\text{S} + 2\text{H}_2\text{CO}_3$ <p>Many metals react with the produced H<sub>2</sub>S to form highly insoluble sulfides:</p> $\text{Fe}^{2+} + \text{H}_2\text{S} \rightarrow \text{FeS (s)} + 2\text{H}^+$ <div>AMD BIOREMEDIATION<ul style="list-style-type: none"><li>Decrease of sulfate concentration</li><li>Removal and recovery of metals</li></ul></div>

SULFIDOGENIC BIOREACTORS		
<p><b>Sulfidogenic bioreactors contain SRB</b>, which treat the entering AMD. They are designed to optimize the production of H<sub>2</sub>S in order to obtain clean process water of neutral pH and low metal concentration. Metals are removed as insoluble sulfides.</p> <p>Two main technologies using sulfidogenic bioreactors have been described: <i>Thiopaq</i> and <i>Biosulfide</i>.</p>	<p><b>Electron donor for SRB</b></p> <p>SRB require the addition of an electron donor to achieve sulfate reduction. To this effort, they use several low molecular weight substrates, such as ethanol and hydrogen.</p> <div><p><b>Ethanol</b></p><ul style="list-style-type: none"><li>» Small-scale applications.</li><li>» High operational costs.</li></ul><math display="block">3\text{SO}_4^{2-} + 2\text{C}_2\text{H}_5\text{OH} \rightarrow 3\text{HS}^- + 3\text{H}_2\text{O} + 3\text{HCO}_3^- + \text{CO}_2</math></div> <div><p><b>Hydrogen gas</b></p><ul style="list-style-type: none"><li>» Large-scale applications.</li><li>» Low production of biomass.</li><li>» Produced on site by reforming natural gas.</li><li>» High investment costs.</li></ul><math display="block">\text{SO}_4^{2-} + 4\text{H}_2 + \text{H}^+ \rightarrow \text{HS}^- + 4\text{H}_2\text{O}</math></div>	<p><b>Reactor type</b></p> <p><b>CIRCOC® gas-lift loop</b></p> <ul style="list-style-type: none"><li>» Optimum mixing and mass transfer of H<sub>2</sub>.</li><li>» 3-phase separator: safeguards biomass retention. Enhance gas/liquid separation.</li><li>» Bacteria grow into aggregates. Excess must be removed.</li></ul>

## THIOPAQ

- » Anaerobic sulfate reduction by SRB and metal precipitation in a single tank ①.
- » (Optional) Aerobic oxidation of any excess H<sub>2</sub>S to elemental sulfur by sulfide-oxidizing bacteria ②.

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graph LR
    AMW[ACID MINE WATER] --> B1[SULFIDOGENIC BIOREACTOR 1]
    ED[ELECTRON DONOR] --> B1
    B1 --> MS[METAL SULFIDES]
    B1 -- "EXCESS H2S" --> B2[BIO-REACTOR WITH SULFUR OXIDIZING BACTERIA 2]
    AIR[AIR] --> B2
    B2 --> S[SULPHUR]
    B2 --> TW[TREATED WATER]
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- *Thiopaq* technology can treat effectively low metal concentration streams.
- However, the process has limitations, especially when it comes to the treatment of high metal concentration streams:
  - » Sensitivity of SRB to low pH and high metal concentration.
  - » Difficulty to maintain the optimum conditions due to seasonal fluctuations of the entering mine water.
  - » Plugging in the bioreactor, as a consequence of precipitation of sulfide sludge.
  - » Sludge contains a mixture of metals.
  - » Sludge also contains biomass.

} Greater volume of sludge increases disposal expense

### Case of Study. Budel Zinc Refinery (Netherlands)

Treatment of AMD by neutralization with lime: production of gypsum

Legislative restrictions for further production of residues

**Thiopaq technology:** treatment of sulfate and recovery of zinc sulfide, which is recycled to the refinery

- » Industrial treatment plant (2000)
- » Influent = 40 m<sup>3</sup>/h
- » Electron donor: hydrogen

Component	Influent (mg/l)	Effluent (mg/l)
Sulfate	15,000	<250
Zinc	10,000	<0.3

## BIOSULFIDE

- » Chemical precipitation of metal sulfides ① separated from biological sulfate reduction ②.
- » In chemical tank, raw AMD contacts with H<sub>2</sub>S generated in the biological tank.

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graph LR
    AMW[ACID MINE WATER] --> MP[METAL PRECIPITATION 1]
    MP --> MS[METAL SULFIDES]
    MP -- "H2S" --> B2[SULFIDOGENIC BIOREACTOR 2]
    ED[ELECTRON DONOR] --> B2
    B2 --> TW[TREATED WATER]
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- *Biosulfide* technology is useful with low and high metal concentration streams.
- It presents some advantageous aspects compared to *Thiopaq* technology:
  - » SRB receive very low concentration of metals.
  - » Optimum conditions easily maintained.
  - » Sulfide sludges isolated in chemical circuit.
  - » Selective separation and removal of metals by using several precipitation tanks.

### Case of study. Kennecott Utah Copper (USA)

#### **Biosulfide technology:**

Treatment of a contaminated groundwater stream with high levels of metals and sulfate.

- » Pilot treatment plant (1995)
- » Influent = 0.2 m<sup>3</sup>/h
- » Electron donor: hydrogen

Component	Influent (mg/l)	Effluent (mg/l)
Sulfate	30,000	<500
Copper	60	<0.1
Iron	675	<0.3
Zinc	65	<0.1
Manganese	350	0.3
Aluminium	2,200	<2

CONCLUSIONS
<ul style="list-style-type: none"><li>» AMD remediation with sulfidogenic reactors can address the threat that mine wastes pose to the environment.</li><li>» In addition to the environmental benefit, it can trigger economic advantages by recovering valuable metals.</li><li>» <i>Biosulfide</i> technology has many upsides in comparison to <i>Thiopaq</i> technology. However, every situation requires an individualized design of the treatment process.</li><li>» There is a lack of awareness in the mining industry and metallurgy in the aspect of implementing active and biological technologies.</li></ul>

REFERENCES
<p>[1] Johnson, D.B., Hallberg, K.B., Acid mine drainage remediation options: a review, <i>Science of the Total Environment</i> (2005), 338 : 3–14.</p> <p>[2] Van Lier, R.J.M., Buisman, C.J.N., Piret, N.L., Thiopaq technology for the mining and metallurgical industries, Paques Bio Systems B. V., The Netherlands.</p> <p>[3] Rowley, M.V., Warkentin, D.D., Yan, V.Y., Piroshco, B.M., The biosulfide process: integrated biological/chemical acid mine drainage treatment - results of laboratory piloting, International Land Reclamation and Mine Drainage Conference, Pittsburgh (1994), 24 – 29.</p>